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But even the humble covalent bond - the basic building block of chemistry - is a consequence of the quantum interactions of electrons. Once these rules were worked out in the 1920s, scientists realised that, for the first time, they had a detailed theory of how chemistry works.

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C_8H_{18} is the formula for octane. The oxygen (O_2) is consumed from the intake air. Nitrogen (N_2) is also present in the atmospheric air, but ideally does not participate in any reactions (it is quite inert at low temperatures). Note that the combustion products are carbon dioxide (CO_2) and water (H_2O), if the combustion is 'perfect'. Also, note that there are the same number of each kind ...

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